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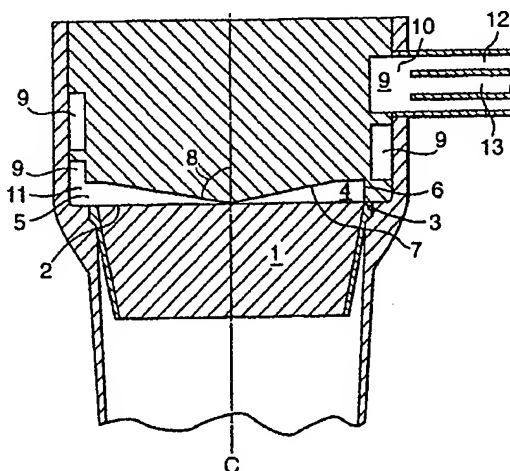
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- (71) Applicant (for all designated States except US): **SHELL INTERNATIONALE RESEARCH MAATSCHAPPIJ B.V.** [NL/NL]; Carel van Bylandtlaan 30, NL-2596 HR The Hague (NL).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): **SEVENHUIJSEN**, Eric [NL/NL]; Badhuisweg 3, NL-1031 CM Amsterdam (NL). **WENTINCK, Hendrik, Martinus** [NL/NL]; Badhuisweg 3, NL-1031 CM Amsterdam (NL).
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(54) Title: FLOW DISTRIBUTOR



(57) Abstract: A reactor comprising a catalyst bed (1) having a central longitudinal axis, which catalyst bed has an upstream surface (2) perpendicular to the central longitudinal axis, and a distribution chamber (4) for directing a gaseous mixture of reactants to the upstream surface (2) of the catalyst bed (1), which distribution chamber (4) has an inlet (5) for tangentially introducing the mixture into the distribution chamber (4), the distribution chamber (4) being defined by the upstream surface (2) of the catalyst bed (1), a side surface (6) having a central longitudinal axis that coincides with the central longitudinal axis of the catalyst bed (1), and a covering surface (7) formed such that the distance between the upstream surface (2) of the catalyst bed (1) and the covering surface (7) is monotonically decreasing towards the central longitudinal axis. The invention further relates to a fuel cell system comprising such a reactor and a fuel cell and to a process for the catalytic oxidation of hydrocarbonaceous fuel using such a reactor.

FLOW DISTRIBUTOR

The present invention relates to a reactor comprising a catalyst bed and a distribution chamber for directing a gaseous mixture of reactants to the upstream surface of the catalyst bed, a fuel cell system comprising such a reactor, and to a process for the catalytic oxidation of a hydrocarbonaceous fuel using such a reactor.

In many conversion processes, reactants are mixed to obtain a gaseous mixture that is contacted with a catalyst bed. For optimal mixing, it is advantageous to use a mixing zone with a relatively large ratio of length to diameter. Therefore, the cross-sectional area of the mixing zone is generally smaller than that of the catalyst bed with which the mixture is to be contacted. In order to distribute the mixture evenly over the upstream surface of the catalyst bed, either a plurality of mixers can be used or the mixture has to be spread over the upstream surface of the catalyst bed.

If the reactants form a mixture that is susceptible to ignition or explosion, such as is the case in processes for oxidation of a hydrocarbonaceous fuel wherein a mixture of hydrocarbonaceous fuel and a molecular oxygen containing gas is contacted with a catalyst, it is important that variations in the residence time of the mixture in the zone upstream of the catalyst are minimal. Zones wherein the flow of the feed mixture is stagnant are to be avoided. Reference herein to zones wherein the flow is stagnant is to zones wherein the residence time of the mixture is relatively long as compared to the average residence time of the mixture.

In EP 303 439, a gas mixer and distributor for feeding a gaseous reaction mixture to a catalytic partial

oxidation reaction zone is disclosed. The disclosed gas mixer and distributor comprises a plurality of small mixing tubes opening above a larger catalytic reaction zone.

5 In WO 98/49095 is disclosed an injector/reactor apparatus for the catalytic partial oxidation of light hydrocarbons comprising a manifold of mixing nozzles upstream of a larger catalytic partial oxidation zone.

10 In WO 98/30322, a catalytic reactor for partial oxidation of a hydrocarbonaceous feedstock comprising a mixer-diffuser for mixing a hydrocarbonaceous feedstock and an oxygen-containing gas and subsequently spreading it over the catalyst surface is disclosed. In the disclosed mixer-diffuser, the mixture is first expanded
15 in an annular chamber by increasing the diameter of the annular flow path of the mixture without increasing the available cross-sectional area of the flow path, the expanded mixture is then supplied via an annular inlet to a distribution chamber. A disadvantage of the mixer-
20 diffuser of WO 98/30322 is that the mixture is axially supplied to the expander/diffuser, thereby requiring a relatively high mixing chamber.

It has now been found that a gaseous mixture of reactants can advantageously be spread over the upstream
25 surface of a catalyst bed by using a reactor wherein the mixture is tangentially introduced into a distribution chamber of which the height is monotonically decreasing towards the central longitudinal axis of the catalyst bed.

30 Accordingly, the present invention relates to a reactor comprising a catalyst bed having a central longitudinal axis, which catalyst bed has an upstream surface perpendicular to the central longitudinal axis, and a distribution chamber for directing a gaseous
35 mixture of reactants to the upstream surface of the

catalyst bed, which distribution chamber has an inlet for tangentially introducing the mixture into the distribution chamber, the distribution chamber being defined by the upstream surface of the catalyst bed, a side surface having a central longitudinal axis that coincides with the central longitudinal axis of the catalyst bed, and a covering surface formed such that the distance between the upstream surface of the catalyst bed and the covering surface is monotonically decreasing towards the central longitudinal axis.

An advantage of the reactor according to the invention is that, during normal operation, the gaseous mixture of reactants flows in a swirling movement, tangentially with respect to the upstream surface of the catalyst, resulting in relatively high strain rates or turbulence such that the risk of flame propagation is low. The monotonically decreasing height of the distribution chamber minimises the occurrence of recirculation of the mixture away from the catalyst surface in the centre of the swirl movement. Another advantage of the reactor according to the invention is that it is suitable to comprise a mixer having a spiral mixing chamber, such that the required length of the mixing chamber may be achieved without the need for a height mixer, resulting in a compact reactor.

The reactor according to the invention will now be further illustrated by way of example with reference to Figures 1 to 3.

Figure 1 shows a longitudinal section of the catalyst bed and distribution chamber of a first embodiment of the reactor according to the invention.

Figure 2 shows a longitudinal section of the catalyst bed and distribution chamber of a second embodiment of the reactor according to the invention.

Figure 3 shows a longitudinal section of the catalyst bed and distribution chamber of a third embodiment of the reactor according to the invention.

5 In Figure 1 is shown a catalyst bed 1 in the form of a frusto-conically shaped, ceramic foam, having a central longitudinal axis C and having its upstream surface 2 perpendicular to the central longitudinal axis C. The upstream surface 2 has a circular outer perimeter 3. Upstream of the catalyst bed 1, the reactor has a
10 distribution chamber 4 for directing a mixture of reactants to the catalyst bed 1.

The distribution chamber 4 has an inlet 5 for tangentially introducing the mixture of reactants into chamber 2. The distribution chamber 4 is defined by the
15 upstream surface 2 of catalyst bed 1, side surface 6, and covering surface 7. The covering surface 7 has a shape such that the height of chamber 4, i.e. the distance between the upstream surface 2 of catalyst bed 1 and the covering surface 7, is monotonically decreasing towards
20 the central longitudinal axis C of catalyst bed 1.

The covering surface 7 can be made of a high-temperature resistant material, preferably a high-alloy steel or a refractory oxide. Part of the covering surface 7 has a conical shape having a half cone angle 8.

25 The reactor in Figure 1 further comprises a helically-wound spiral mixing channel 9, having an inlet 10 for the reactants to be mixed and an outlet 11 for the mixture of reactants. The inlet 10 is in fluid communication with inlet conduits 12 and 13.
30 Alternatively, channel 9 may have separate inlets, each in fluid communication with a single inlet conduit. The outlet 11 is in fluid communication with the inlet 5 of the distribution chamber 4. The channel 9 may be provided with obstacles (not shown) to promote mixing.

The reactor shown in Figure 2 comprises an inlet channel 14 for supplying part of the mixture of reactants to the upstream surface 2 of the catalyst bed 1, the inlet channel 14 having an outlet 15 opening into the covering surface 7. Part of the mixture of reactants is fed to the inlet channel 14 via inlet 16. In the embodiment shown in Figure 2, inlet 16 is in fluid communication with the mixing channel 9.

Preferably, inlet channel 14 contains an igniter 17 for igniting the mixture during start-up of the reactor. Suitable igniters are known in the art.

In the embodiment of the reactor shown in Figure 3, the catalyst bed 1 is an annular catalyst having an upstream surface 2 with an outer perimeter 3 and an inner perimeter 18. The reactor of Figure 3 further comprises a conduit 19 for removal of the effluent of the catalyst bed 1. The annular catalyst bed 1 is arranged around conduit 19. This embodiment is advantageous in situations wherein it is desired to supply the feed mixture and remove the effluent of catalyst bed 1 at the same side of the catalyst bed 1.

Part of the covering surface 7 has a frusto-conical shape having a half cone angle 8.

The reactor in Figure 3 comprises a circularly-wound spiral mixing channel 9.

In order to create a tangentially flowing feed mixture in the reactor according to the invention, the inlet 5 for introducing the mixture into distribution chamber 4 is preferably located in side surface 6.

The upstream surface 2 preferably has a circular or oval outer perimeter 3, preferably a circular outer perimeter. Alternatively, the outer perimeter 3 of the upstream surface 2 is a polygonal having more than four angles.

If the catalyst bed 1 is an annular catalyst bed, the distribution chamber 4 is an annular chamber. Since in such a chamber the space in the centre of the swirl movement is not available to the gaseous mixture flowing in the chamber, there is no need for a monotonically increasing height in order to prevent recirculation in the centre of the swirl movement. Therefore, the height of the distribution chamber 4, i.e. the distance between the upstream surface 2 and the covering surface 7, may be constant in the case of an annular catalyst bed.

Preferably, at least part of the covering surface 7 has a conical shape, such as shown in Figure 1, or, in the case of an annular catalyst bed 1, as illustrated in Figure 3, a frusto-conical shape. The half cone angle θ is preferably larger than 45° , more preferably larger than 60° , even more preferably larger than 75° . A conically or frusto-conically shaped covering surface 7 with a relatively large half cone angle minimises the occurrence of dead spaces in the distribution chamber 4 and limits the height of the chamber 4 at the outer perimeter 3 as compared to the diameter of the upstream surface 2.

Preferably, the diameter of the upstream surface 2 is at least 5 times the largest height of the distribution chamber 4, i.e. the distance between the upstream surface 2 and the covering surface 7 at the outer perimeter 3. More preferably, the diameter of the upstream surface 2 is at least 10 times the largest height of chamber 4, even more preferably at least 15 times. By limiting the height of the distribution chamber, a sufficient high flow velocity component parallel to the catalyst surface can be maintained. If the upstream surface has not a circular outer perimeter, reference to the diameter is to the average width of the upstream surface.

Preferably, the smallest distance between the upstream surface 2 and the covering surface 7 is less than 1 mm, more preferably less than 0.5 mm. It is most preferred that the covering surface 7 approaches or touches the upstream surface, such that the smallest distance between the upstream surface 2 and the covering surface 7 is substantially zero.

Reference herein to a catalyst bed is to a fixed arrangement of catalyst, such as a fixed bed of catalyst particles, a metal or ceramic monolithic structure provided with catalytically active material, or an arrangement of metal wires or gauzes provided with catalytically active material.

Preferably, the catalyst bed is suitable for the partial oxidation of a hydrocarbonaceous fuel. Catalytic partial oxidation processes and suitable catalysts therefore are known in the art, for example from US 5,149,464, EP 576 096, EP 629 578, WO 99/37580 and WO 99/19249. Such catalysts generally comprise, as catalytically active component, a metal selected from Group VIII of the Periodic Table of the Elements. Catalysts comprising, as the catalytically active component, a metal selected from rhodium, iridium, palladium and platinum are preferred. Catalysts comprising rhodium and/or iridium are most preferred.

The catalytically active metal is most suitably supported on a carrier, such as refractory oxide particles, monolith structures, or metallic arrangements such as metal gauzes or arrangements of metal wires. Suitable carrier materials are well known in the art and include refractory oxides, such as silica, alumina, titania, zirconia and mixtures thereof, and metals, such as aluminium-containing high-temperature resistant alloys.

Typically, the partial oxidation catalyst comprises the active metal in an amount in the range of from 0.01 to 20% by weight, based on the weight of carrier material, preferably from 0.02 to 10% by weight, more preferably from 0.1 to 7.5% by weight.

In a specific embodiment of the reactor according to the invention, the reactor comprises, in addition to a partial oxidation catalyst bed as hereinbefore defined, a catalytic reaction zone for the water-gas shift conversion of the effluent of the partial oxidation catalyst bed. Optionally, the reactor further comprises a zone for the catalytic selective oxidation of the remaining carbon monoxide in the effluent of the catalytic reaction zone for the water-gas shift conversion.

Such a reactor comprising in series a partial oxidation catalyst bed, a catalytic water-gas shift reaction zone and, optionally, a zone for the catalytic selective oxidation of carbon monoxide may be advantageously applied for converting a hydrocarbonaceous fuel into a hydrogen-rich gas to be further processed in a fuel cell.

The invention further relates to a fuel cell system comprising the reactor as hereinbefore defined, wherein the catalyst bed is a catalyst bed for the partial oxidation and a hydrocarbonaceous fuel. The reactor of the fuel cell system according to the invention may additionally comprise a catalytic water-gas shift reaction zone and, optionally, a zone for the catalytic selective oxidation of carbon monoxide as hereinbefore described. The fuel cell system according to the invention may further comprise a catalytic after burner for combusting the effluent gases from the fuel cell.

Since the shape of the distribution chamber is such that re-circulation of the feed mixture is minimised, the

reactor according to the invention is particularly suitable for those processes wherein reactants that form a mixture that is susceptible to ignition and explosion have to be mixed prior to contacting them with the catalyst bed. This is for example the case in the catalytic oxidation of a hydrocarbonaceous fuel.

Accordingly, the present invention also relates to a process for the oxidation of a hydrocarbonaceous fuel, using the reactor as hereinbefore defined. Preferably, the process is a process for the catalytic partial oxidation of a hydrocarbonaceous fuel.

Catalytic partial oxidation processes are typically carried out by contacting a feed mixture comprising a hydrocarbonaceous fuel and an oxygen-containing gas with a suitable catalyst, preferably in amounts giving an oxygen/carbon ratio in the range of from 0.3 to 0.8, more preferably of from 0.45 to 0.75, even more preferably of from 0.45 to 0.65. References herein to oxygen/carbon ratio are to the ratio of oxygen in the form of molecules O_2 to carbon present in the fuel.

The feed mixture may comprise steam and/or carbon dioxide. If steam is present in the feed mixture, the steam/carbon ratio, i.e. the ratio of molecules of steam (H_2O) to carbon in the fuel, is preferably in the range of from above 0.0 to 3.0, more preferably of from above 0.0 to 2.0.

The hydrocarbonaceous fuel may comprise hydrocarbons and/or oxygenates which are gaseous at the temperature and pressure prevailing in the catalyst bed during normal operation of the catalytic partial oxidation process. Particular suitable fuels comprise hydrocarbons which are gaseous or liquid under standard conditions of temperature and pressure (STP, i.e. at 0 °C and 1 atmosphere), such as natural gas, naphtha, kerosene, or gas oil. Reference herein to oxygenates is to molecules

containing, apart from carbon and hydrogen atoms, at least one oxygen atom which is linked to either one or two carbon atoms or to a carbon atom and a hydrogen atom, such as alcohols, ethers, acids and esters.

5 Reference herein to an oxygen-containing gas is to a gas containing molecular oxygen, for example air, substantially pure oxygen or oxygen-enriched air. The oxygen-containing gas preferably is air.

10 The temperature prevailing in a catalytic partial oxidation process is typically in the range of from 750 to 1500 °C, preferably of from 800 to 1350 °C, more preferably of from 850 to 1300 °C. Reference herein to temperature is to the temperature in the upstream layer of the catalyst bed.

15 The catalytic partial oxidation process is typically operated at a pressure in the range of from 1 to 100 bar (absolute), preferably in the range of from 1 to 50 bar (absolute), more preferably of from 1 to 10 bar (absolute).

20 The feed mixture is preferably contacted with the catalyst bed at gas hourly space velocities which are in the range of from 10,000 to 10,000,000 Nl/kg/h, more preferably in the range from 20,000 to 5,000,000 Nl/kg/h, even more preferably in the range of from 50,000 to
25 3,000,000 Nl/kg/h. Reference herein to normal litres is to litres at STP.

The invention will now be further illustrated by means of the following example.

EXAMPLE

30 A stream of natural gas (240 g/h) was mixed with a stream of air (930 g/h) in a reactor as shown in Figure 3. The diameter of the upstream surface of catalyst bed 1 was 80 mm and the height of distribution chamber 3 at outer perimeter 18 was 5 mm. Both streams
35 were pre-heated to give a temperature between 350 and

400 °C at inlet 10. The mixture was contacted with the catalyst bed 1 and converted into a mixture comprising carbon monoxide and hydrogen. The operating pressure was 1 bar (absolute). The temperature at the wall of mixing channel 8 at the location indicated with T was measured by means of a thermocouple. The measured temperature was 600 °C. No ignition or explosion occurred.

The catalyst bed 1 was a 65 ppi (pores per inch) foam of zirconia partially-stabilised with yttria (Y-PSZ) provided with 2.5 wt% Rh, 2.5 wt% Ir, and 7 wt% Zr based on the total weight of catalyst by impregnating the foam with an aqueous solution comprising rhodium trichloride, iridium tetrachloride and zirconyl nitrate, followed by drying and calcining.

C L A I M S

1. A reactor comprising a catalyst bed having a central longitudinal axis, which catalyst bed has an upstream surface perpendicular to the central longitudinal axis, and a distribution chamber for directing a gaseous mixture of reactants to the upstream surface of the catalyst bed, which distribution chamber has an inlet for tangentially introducing the mixture into the distribution chamber, the distribution chamber being defined by the upstream surface of the catalyst bed, a side surface having a central longitudinal axis that coincides with the central longitudinal axis of the catalyst bed, and a covering surface formed such that the distance between the upstream surface of the catalyst bed and the covering surface is monotonically decreasing towards the central longitudinal axis.
2. A reactor according to claim 1, wherein at least part of the covering surface has a conical shape.
3. A reactor according to any of claims 1, wherein the catalyst bed is an annular catalyst bed.
4. A reactor according to claim 3, wherein at least part of the covering surface has a frusto-conical shape.
5. A reactor according to claim 2 or 4, wherein the cone half angle of the covering surface is at least 45°, preferably at least 60°, more preferably at least 75°.
6. A reactor comprising an annular catalyst bed having a central longitudinal axis, which catalyst bed has an upstream surface perpendicular to the central longitudinal axis, and an annular distribution chamber for directing a gaseous mixture of reactants to the upstream surface of the catalyst bed, which distribution chamber has an inlet for tangentially introducing the

mixture into the distribution chamber, the distribution chamber being defined by the upstream surface of the catalyst bed, a side surface having a central longitudinal axis that coincides with the central longitudinal axis of the catalyst bed, and a covering surface formed such that the distance between the upstream surface of the catalyst bed and the covering surface is constant.

7. A reactor according to any of claims 3 to 6, further comprising a conduit for removal of effluent of the catalyst bed, wherein the annular catalyst bed is arranged around the conduit.

8. A reactor according to any of claims 1 to 5, and 7, wherein the smallest distance between the upstream surface and the covering surface is less than 1 mm, preferably less than 0.5 mm, more preferably is substantially zero.

9. A reactor according to any of the preceding claims, further comprising a spiral mixing channel for mixing reactants to obtain the gaseous mixture of reactants, having an inlet for the reactants and an outlet, wherein the outlet of the mixing channel is in fluid communication with the inlet of the distribution chamber.

10. A reactor according to any of the preceding claims, wherein the ratio of the diameter of the upstream surface to the largest distance between the upstream surface and the covering surface is at least 5, preferably at least 10, more preferably at least 15.

11. A reactor according to any of the preceding claims, further comprising an inlet channel for supplying part of the mixture of reactants to the upstream surface of the catalyst bed, the inlet channel having an outlet opening into the covering surface.

12. A reactor according to claim 11, wherein the inlet channel contains an igniter.

13. A reactor according to any of the preceding claims, wherein the catalyst bed comprises a catalyst for the partial oxidation of a hydrocarbonaceous fuel.
- 5 14. A reactor according to claim 13, further comprising a catalytic reaction zone for water-gas shift conversion of the effluent of the catalyst bed and, optionally, a catalytic reaction zone for selective oxidation of carbon monoxide in the effluent of the catalytic reaction zone for water-gas shift conversion.
- 10 15. A fuel cell system comprising a reactor according to claim 13 or 14 and a fuel cell.
16. A process for the catalytic oxidation of a hydrocarbonaceous fuel, using the reactor according to any of claims 1 to 13, preferably for the catalytic
- 15 partial oxidation of a hydrocarbonaceous fuel.

Fig.1.

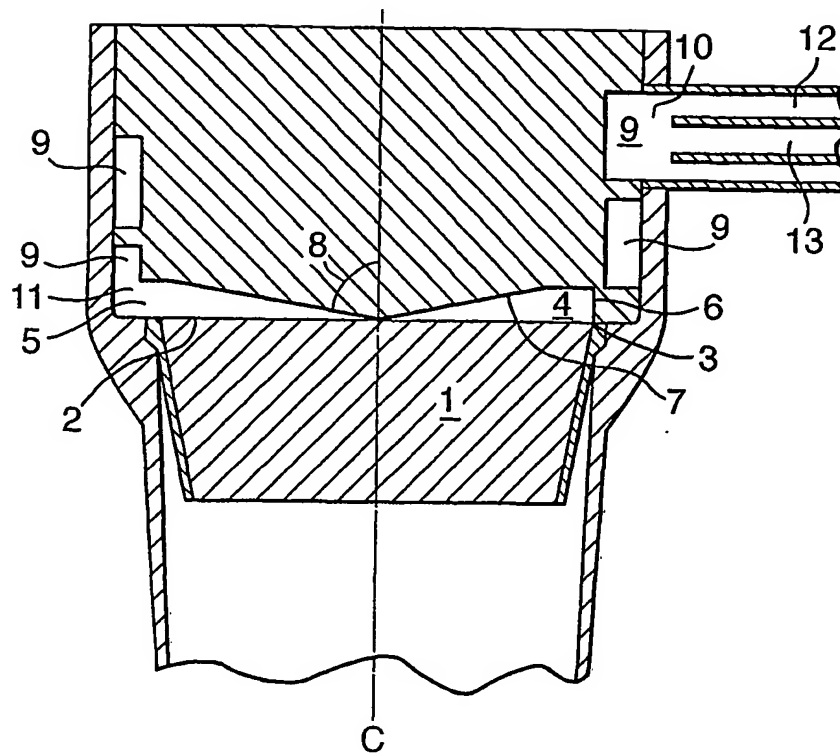
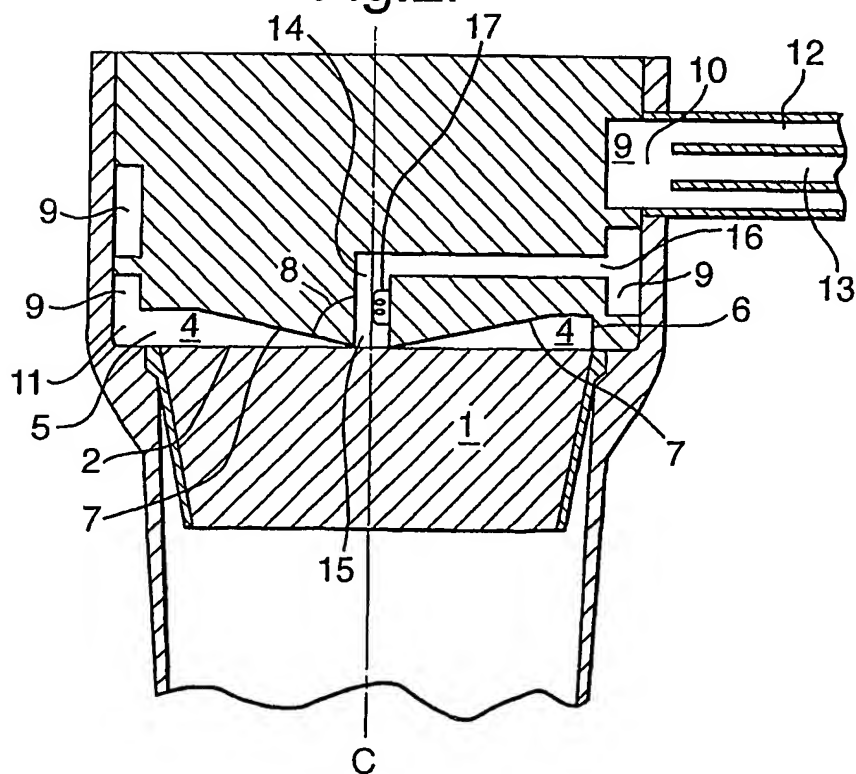


Fig.2.



INTERNATIONAL SEARCH REPORT

International Application No
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A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 B01J8/02 C01B3/38 C01B3/48 H01M8/06

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B01J C01B H01M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	FR 1 177 434 A (HEINRICH KOPPERS GMBH) 24 April 1959 (1959-04-24) the whole document	1,2
A	US 5 873 929 A (ANDREANI PHILIPPE, MONEREAU CHRISTIAN, POTEAU MICHEL) 23 February 1999 (1999-02-23) column 2, line 8 - line 58 column 3, line 7 - line 29 figure 1	1,2,5
A	US 5 779 773 A (CAM FRANCOIS, PHELUT SERGE) 14 July 1998 (1998-07-14) column 4, line 61 - column 5, line 52 column 6, line 56 - line 61 claims 1-17; figures 1,7-11	1,9
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☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3018

Authorized officer

Vlassis, M

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 01/09101

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 463 247 A (UDP INC) 2 January 1992 (1992-01-02) column 7, line 43 -column 9, line 31 figures 1,2 ----	1,2
A	DD 234 562 A (VEB GERMANIA KARL-MARX-STADT) 9 April 1986 (1986-04-09) page 4, paragraph 3 - paragraph 8 claims 1-4; figures 1,2 ----	1
A	DE 43 03 471 A (NOELL DBI ENERGIE ENTSORGUNG) 11 August 1994 (1994-08-11) the whole document -----	6,7

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 01/09101

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
FR 1177434	A	24-04-1959	NONE	
US 5873929	A	23-02-1999	FR 2750617 A1	09-01-1998
			BR 9706553 A	20-07-1999
			CN 1197412 A	28-10-1998
			EP 0815932 A1	07-01-1998
			WO 9800230 A1	08-01-1998
			JP 11514074 T	30-11-1999
US 5779773	A	14-07-1998	FR 2730423 A1	14-08-1996
			BE 1010543 A3	06-10-1998
			DE 19605257 A1	22-08-1996
			GB 2297923 A ,B	21-08-1996
			GB 2329597 A ,B	31-03-1999
EP 0463247	A	02-01-1992	US 4938422 A	03-07-1990
			AU 631658 B2	03-12-1992
			AU 5794990 A	02-01-1992
			EP 0463247 A1	02-01-1992
			US 5098690 A	24-03-1992
DD 234562	A	09-04-1986	DD 234562 A3	09-04-1986
DE 4303471	A	11-08-1994	DE 4303471 A1	11-08-1994